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Isomerization Polymerization of 4-Alkylcyclopentenes Catalyzed by Pd Complexes: Hydrocarbon Polymers with Isotactic-Type Stereochemistry and Liquid-Crystalline Properties

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The physical properties of $poly(\alpha$ -olefins) vary to a large degree depending on the alkyl substituents derived from the monomer and the relative stereochemistry of the neighboring repeating units (tacticity). Polymers with five- or six-membered rings in the monomer units have unique thermal and optical properties.¹ Early transition-metal complex-catalyzed polymerization of cyclic olefins and cyclopolymerization of dienes provide these polymers efficiently, although polymerization with regulated tacticity is rare.² Ni- or Pd-diimine complexes3 catalyze the polymerization of α -olefins and cyclopolymerization of dienes with isotactic selectivity⁴ when a C_2 -symmetric ligand is employed. The polymerization reactions, including cyclopolymerization of monoterminal dienes,⁵ involve isomerization of the growing polymer end via a chainwalking process. In this paper, we report polymerization of 4-alkylcyclopentenes catalyzed by Pd complexes to afford polymers with 1,3-disubstituted five-membered rings formed via olefin insertion and chain walking.^{6,7} The polymer with regulated tacticity exhibited liquid-crystalline properties.8

Pd-dimine complexes **1a** and **1b**, with C_{2V} and C_2 -symmetric ligands,⁵ respectively, promote the polymerization of 4-methylcyclopentene (**I-1**) in the presence of NaBARF (BARF = [{3,5-(CF_3)_2C_6H_3}_4B]⁻) to produce poly(**I-1**) (eq 1):



The ¹³C{¹H} NMR spectrum of poly(**I-1**) obtained using **1a**/NaBARF [Figure 1(i)] exhibits three peaks (39.32, 39.26, and 39.22 ppm)⁹ for the CH₂ (b₁) carbon between two CH carbons, which corresponds to the triad structures with respect to the orientation of three neighboring five-membered rings. The ratio among the peaks (31:47:22) suggests a random or atactic-type arrangement of the five-membered rings. Poly(**I-1**) obtained using **1b**/NaBARF, on the other hand, gives rise primarily to the signal at the lowest field position (39.32 ppm) [Figure 1(ii)]. It is assigned to three monomer units with the same stereochemistry, an *rr* triad, indicating a highly threo-diisotactic structure of the polymer (*rr* > 93%).

The different tacticities of poly(I-1) formed by different catalysts affected the crystallographic results. XRD measurements (Cu K α)¹⁰ on the polymer obtained using 1a/NaBARF showed an intense



Figure 1. ${}^{13}C{}^{1H}$ NMR spectra (125 MHz, CDCl₃ at 50 °C) of (i) poly(I-1) (run 1 in Table 1), (ii) poly(I-1) (run 2), (iii) poly(I-3) (run 3), (iv) poly(I-4) (run 4), and (v) poly(I-10) (run 5).

diffraction peak at $2\theta = 17.92^{\circ}$ and a halo pattern, indicating the coexistence of the crystalline and amorphous domains. The isotactic polymer obtained using **1b**/NaBARF exhibited only a single sharp diffraction peak at $2\theta = 18.42^{\circ}$ and thus contained the crystalline domain almost exclusively at room temperature.

Table 1. Polymerization of 4-Alkylcyclopentenes^a

run	olefin	Pd	time	conv. (%)	Mn	M _w /M _n
1	I-1	1a	3 h	quant.	$10100^{b,c}$	$1.90^{b,c}$
2	I-1	1b	5 min	quant.	$8200^{c,d}$	-
3	I-3	1b	1 h	quant.	3000 ^e	2.18^{e}
4	I-4	1b	1.5 h	66	3000 ^e	3.38 ^e
5	I-10	1b	36 h	56	5700 ^e	2.38 ^e

^{*a*} Reaction conditions: Pd complex (0.010 mmol), NaBARF (0.012 mmol), [monomer]/[Pd] = 300, CH₂Cl₂ solvent, r.t. ^{*b*} GPC (CHCl₃). ^{*c*} GPC (*o*-dichlorobenzene) measurements on poly(**I**-**1**) (runs 1 and 2) were not conducted because of low solubility. ^{*d*} Estimated by ¹H NMR. ^{*e*} GPC (*o*-dichlorobenzene).

Scheme 1 summarizes a plausible mechanism for the polymerization of **I-1**. Coordination and insertion of the monomer produce a cyclopentylpalladium intermediate (**A**). It does not undergo direct insertion of the monomer into the Pd—secondary alkyl bond but causes isomerization via chain walking to produce the cyclopen-

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tylmethyl palladium species (**B**).³ Insertion of a new monomer into the Pd-primary alkyl bond leads to further polymer growth. Formation of a trans five-membered ring in the polymer is explained by assuming preferential coordination of the monomer to Pd at the less-hindered side followed by smooth insertion and chain walking.

Scheme 1. Proposed Mechanism for the Polymerization of I-1

$$H \xrightarrow{CH_3} CH_3 \longrightarrow H \xrightarrow{Pd} CH_3 \longrightarrow H \xrightarrow{Pd} CH_3 \longrightarrow H \xrightarrow{Pd} CH_3$$

$$A$$

$$H \xrightarrow{CH_2 \sim Pd} \longrightarrow (f_2 \sim P_3)$$

$$B$$

Scheme 2 depicts a possible coordination mode of the monomer to the Pd center of 1b activated by NaBARF. Steric repulsion between the bulky naphthyl substituent and the monomer renders coordination in C more favorable than that in D. Repetitive coordination in C and insertion of the coordinated monomer would result in formation of the isotactic polymer.

Scheme 2. Selective Coordination of I-1



Polymerization of 4-alkylcyclopentenes with longer alkyl groups (I-3, I-4, and I-10) catalyzed by 1b/NaBARF afforded the corresponding polymers poly(I-3), poly(I-4), and poly(I-10) (eq 1 and runs 3-5 in Table 1). On the basis of the ${}^{13}C{}^{1}H$ NMR results [Figure 1(iii)–(v)], the polymers contain *trans*-fused 1,3-cyclopentane rings located at regulated intervals along the linear chain. The long spacers between the five-membered rings did not allow estimation of the tacticity of these polymers by NMR analysis. The polymerization of the monomers with more bulky substituents than I-1 using the C_2 -symmetric catalyst 1b probably yields poly(I-3), poly(I-4), and poly(I-10) with the regulated tacticity.

Differential scanning calorimetry (DSC) measurements on isotactic poly(I-1) and poly(I-3) showed monotropic transitions, while poly(I-4) and poly(I-10) with longer alkylene spacers undergo enantiotropic transitions. Increasing the length of the alkyl spacer leads to lower phase-transition temperatures.¹⁰ These polymers are considered to show liquid-crystalline properties between the transition temperatures. Direct observation of poly(I-4) by polarizing optical microscopy also supports the existence of the liquidcrystalline phase (Figure 2). Poly(I-1) obtained using 1a/NaBARF $(M_{\rm n} = 10100 \text{ and } 4500)$, however, undergoes only an irreversible glass transition and does not show liquid-crystalline behavior. Recently, Naga et al.⁸ reported the formation of liquid-crystalline hydrocarbon polymers having five-membered rings from the cyclopolymerization of 1,5-hexadiene, although the stereochemical structure of the polymer has not been clarified and is probably not regulated.



Figure 2. Polarizing optical micrograph of poly(I-4) (run 4 in Table 1).

In summary, the polymerization of 4-alkylcyclopentenes catalyzed by Pd complexes affords polymers with *trans*-1,3-disubstituted cyclopentane rings located at regulated intervals along the linear polymer chain. The C_2 -symmetric catalyst produced polymers with an isotactic structure that exhibit high crystallinity at low temperature and liquid-crystalline properties upon heating.

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Supporting Information Available: Experimental procedures and NMR spectra, DSC profiles, polarized optical micrographs, and XRD data of polymers. This material is available free of charge via the Internet at http://pubs.acs.org.

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- (9) Determined from the ${}^{13}C{}^{1}H$ NMR spectrum at 230 MHz.
- (10) The Supporting Information contains complete diffraction and DSC data.

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